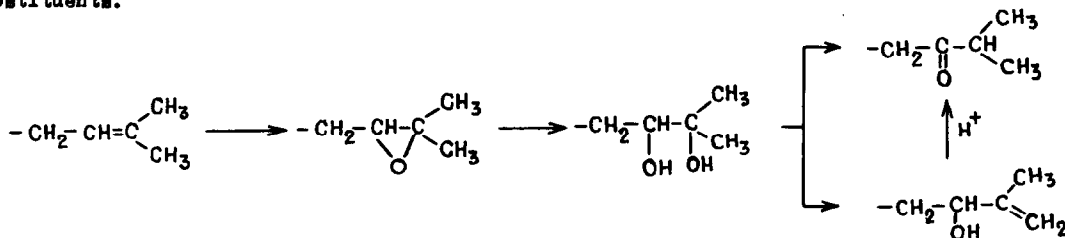


PABULENOL, A BIOLOGICAL TRANSFORMATION PRODUCT OF OXYPEUCEDANIN

S. C. Basa, J. Chatterjee and Mrs. A. Chatterjee
University College of Science, Calcutta-9 (India)

(Received in UK 26 April 1971; accepted in UK for publication 29 April 1971)

The simultaneous occurrence of the furocoumarins, isosimperatorin (I), oxypeucedanin (II), oxypeucedanin hydrate (III), isooxypeucedanin (IV)¹⁻² and pabulenol (V) in Frangos pabularia Lindl. (Umbelliferae) provides circumstantial evidence in support of the following sequence of cellular reactions as visualised by Hendrickson and Richards³ in their biogenetic codification of isoprenoid substituents.



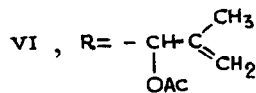
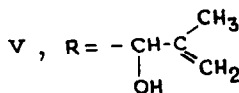
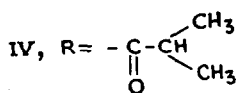
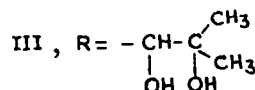
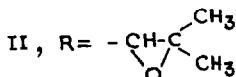
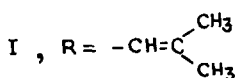
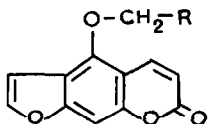
Of these pabulenol (V), $C_{16}H_{14}O_5$ ($M^+286.0852$), m.p. 134-35°, $[\alpha]_D^{25} -3.8^\circ$ (EtOH), is a new coumarin isolated in 0.005% yield from the root portion of P. pabularia. The present communication deals with the structure elucidation of this compound. Its UV spectrum in the region 225-340 nm $[\lambda_{max}^{EtOH} 249.5, 267 \text{ and } 308 \text{ nm (log } \epsilon : 4.17, 4.12 \text{ and } 4.02 \text{ respectively)}]$ is typical of a linear furocoumarin bearing similarity with bergapten^{4,5} while the IR spectrum revealed a hydroxyl group (2.86 μ), a lactone carbonyl (5.85 μ), a benzofuran (9.3 μ), an aromatic ether (9.05 and 8.24 μ) and a terminal methylene (11.2 μ). The 100 MHz NMR spectrum (d_6 -DMSO) showed the coumarin doublets at δ 6.25 and 8.25 (J=10 Hz), the α -furan proton at δ 7.95 [1H(d), J=2 Hz] while the β -furan merged with the single aromatic proton resulting in a finely split two proton signal at δ 7.24 (J=1 Hz). The three proton multiplet at δ 4.35 has been assigned to the two methylene and the adjacent methine protons ($OCH_2-CH-O-$) and the signal at δ 5.36 [1H(d), J=4 Hz] disappearing in D_2O , to the hydroxyl proton. A three proton singlet at δ 1.71 has been attributed to the vinyl methyl and the signals [1H(s) each] at δ 5.06 and 4.86 to $-CH_2$.

The presence of the hydroxyl group as $-CHOH$ was corroborated from the formation of pabulenol acetate (VI), $C_{18}H_{16}O_6$ (M^+328), m.p. 119.5°, and from the study of its NMR spectrum.

The signal at δ 2.12 [$\overline{3H}$ (s)] was assigned to the acetoxy function. The carbinol methine proton now appeared downfield at δ 5.65 [$\overline{1H}$ (t), $J = 5$ Hz] while the β -furan proton resonated as a doublet at δ 6.96 ($J = 3$ Hz). The latter was further split by long range coupling with the aromatic proton at δ 7.18 ($J = 1$ Hz), a characteristic feature of bergaptol derivatives⁶. This demanded the formulation of the side chain in pabulenol acetate as depicted in (VI).

Acid hydrolysis (10% H_2SO_4) of pabulenol afforded isooxypeucedanin (IV) and oxypeucedanin hydrate (III), thereby confirming the presence of an allylic hydroxyl group and a terminal methylene function. Isooxypeucedanin (IV) was obtained from (II) in better yield (60%) by acid catalysed rearrangement with BF_3 , compared to the older method using mineral acids^{1,7-8} (6%).

All the above data can best be reconciled in structure (V) for pabulenol.



Acknowledgements

Sincere thanks are accorded to Dr. G. F. Smith (Manchester) and to Dr. B. C. Das (France) for spectral measurements and to the CSIR (India) for financial assistance.

References

1. G. A. Kusnetsova, *Natural Coumarins and Furocoumarins*, 1967, p. 116-119, Nauka (USSR).
2. S. C. Basa, J. Chatterjee and A. Chatterjee, *Chem. & Ind.*, 746 (1970).
3. J. H. Richards and J. B. Hendrickson, *The Biosynthesis of Steroids, Terpenes and Acetogenins*, 1964, p.110, W. A. Benjamin (Inc. New York).
4. A. G. Caldwell and E. R. H. Jones, *J. Chem. Soc.*, 541 (1945).
5. D. P. Chakraborty and S. K. Chakraborty, *The Ultraviolet Absorption Spectra of some Natural Coumarins*, *Trans. Bose Research Institute*, 24, 15 (1961).
6. F. N. Lahay and J. K. McLeod, *Tet. Lett.*, 447 (1968).
7. E. Späth and K. Klager, *Ber. dtsch. Chem. Ges.*, 66, 914 (1933).
8. H. Böhme and G. Pietsch, *Ber. dtsch. Chem. Ges.*, 72, 773 (1939).